

Pogil Chemistry Polyatomic Answer Key

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~~Answers will vary. There are two extra electrons in the ion compared to a neutral S atom. There are two more electrons than protons in the S2- ion. Two electrons have been added to an atom of sulfur. 6. The dotted line around each polyatomic ion in Model 1 shows that the group of atoms has a charge.~~

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~~Access PDF Polyatomic Ions Pogil Answer Key. group of atoms that are covalently bonded together and which carry an overall electrical charge. The ammonium ion, NH 4 +, is formed when a hydrogen ion (H +) attaches to the lone pair of an ammonia (NH 3) molecule in a coordinate covalent bond. Figure 1.~~

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~~O O. 1+. N H H H H H N S O O O. | | 2 | |. 2 POGIL Activities for High School Chemistry 2. Consider the terms [monatomic] and [polyatomic] as they are used in Model 1. Write a defini- tion for each of these terms. It may be helpful to break the words apart (i.e. poly = atomic). Make sure your group comes to consensus.~~

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~~Chemical Bonding POGIL Activity 7 | A Mixed Up Activity Answer Key # 1 & #2. Chemical Formula Compound Name Monatomic Ions (#1) Polyatomic Ions (#2) CaSO 4 calcium sulfate calcium, Ca +2 sulfate, SO 4 -2 CaSO 3 calcium sulfite calcium, Ca +2 sulfite, SO 3 -2 Na 3 PO 4 sodium phosphate sodium, Na +1 phosphate, PO 4 -3 Li 2 CO 3 lithium carbonate lithium, Li +1 carbonate, CO 3 -2 NH 4 Cl ammonium chloride chloride, Cl -1 ammonium, NH 4 +1 Be (NO 2) 2 beryllium nitrite beryllium, Be +2 ...~~

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~~Pogil: Polyatomic Ion Key Based on your knowledge of monatomic ions, propose an explanation for the net charge on a polyatomic ion. Your answer should include a discussion of subatomic particles.~~

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