

## Pogil Oxidation And Reduction Answer Key

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### NAME: AP Chemistry DATE: POGIL: Oxidation and Reduction

The process of oxidation and reduction can be thought of as a transfer of electrons from one atom to another. Thus, one atom gives up electrons and the other atom gains them. As a result of this process, the oxidation numbers of both atoms change.

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### Answers To Oxidation And Reduction Pogil Activity

Download Free Pogil Oxidation And Reduction Answer Key. oxidation numbers to each species in the equation.  $Cu+2N+5O-2+ AlO\cancel{E} Al+3N+5O-2+ Cu0$ . Step 2: Identify the species with an increase in charge....

### Pogil Answer Key Oxidation And Reduction

April 15th, 2019 - Oxidation Reduction Reactions Answer Key 4 51 If nitric acid is a strong oxidizing agent and zinc is a strong reducing agent then zinc metal will probably reduce nitric acid when the two react that is N will gain electrons and the oxidation number of N must decrease DOWNLOAD 35 POGIL REDOX ANSWER KEY mitfive org

### Oxidation and reduction pogil key

Online Library Pogil Oxidation And Reduction Answer Key Oxidation And Reduction Pogil Answers | Answers Fanatic Oxidation and reduction always involve transfer of electrons. Therefore, there is never oxidation without reduction and vice versa in a redox reaction. Oxidizing something must cause something else to be reduced and vice versa. Therefore, Page 7/22

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Answer Key Oxidation And Reduction Pogil Ws Answers, Keq Lab; Lab... Step 1: Assign oxidation numbers to each species in the equation.  $Cu+2N+5O-2+ AlO\cancel{E} Al+3N+5O-2+ Cu0$ . Step 2: Identify the species with an increase in charge.  $Al0$  (on left side of equation) increases to  $Al+3$ (on right side of equation) Oxidation Half-reaction:  $AlO\cancel{E} Al+3+ 3e^-$ .

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from Model 1 to support your answer. 2 POGIL ... The process of oxidation and reduction can be thought of as a transfer of electrons from one atom to another. Thus, one atom gives up electrons and the other atom gains them. As a result of this process, the

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