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Solution The general solution method is best explained in terms of the complex reaction network notation where the rate for the  $j$ th species is given :  $N_k \frac{dy_j}{dt} = \sum_{l=1}^L \nu_{jl} r_l$  (1) A solution is assumed of the form:  $y_j = Y_j e^{-\lambda t}$  (2) where  $Y_j$  and  $\lambda$  are constants to be determined.

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Elements of Chemical Reaction Engineering

So that it is easier to visualize the

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solution, we first plot the inverse of the reaction rate versus conversion. This type of plot is often called a "Levenspiel Plot.":  
Recalling the mole balance equations for a CSTR and a PFR: Until the conversion ( $X$ ) reaches 0.5, the reaction rate is independent of conversion and the reactor volumes will be ...

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